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## Key indicators

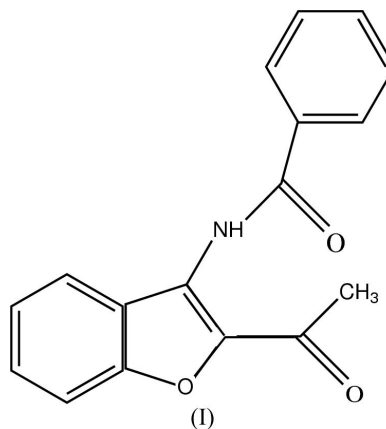
Single-crystal X-ray study  
 $T = 293$  K  
Mean  $\sigma(\text{C}-\text{C}) = 0.007$  Å  
 $R$  factor = 0.073  
 $wR$  factor = 0.195  
Data-to-parameter ratio = 13.7For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

## 2-Acetyl-3-(benzoylamino)-1-benzofuran

The title compound,  $\text{C}_{17}\text{H}_{13}\text{NO}_3$ , displays the characteristic features of benzofuran derivatives. The molecule is nearly planar. An intramolecular  $\text{N}-\text{H}\cdots\text{O}$  hydrogen bond, together with  $\pi-\pi$  stacking interactions, helps to stabilize the structure.

## Comment

The 2- and 3-substituted benzofuran nucleus is a central component of a diverse class of heterocyclic natural and synthetic products that exhibit a broad range of biological activities (Nicolaou *et al.*, 2000; Gölitzer & Kramer, 2000). The title compound, (I), was synthesized from the reaction of 2-acetyl-3-aminobenzofuran (Ocak Iskeleli *et al.*, 2005) with benzoyl chloride in dry acetone.



The benzofuran ring system of the title compound, (I), is planar, with a maximum deviation from the plane of 0.050 (6) Å for atom C4 (Fig. 1). The dihedral angle between the benzofuran ring system and the phenyl ring is 2.47 (13)°. The  $\text{C}=\text{O}$  bond lengths [ $\text{C11}-\text{O3} = 1.232$  (5) Å and  $\text{C9}-\text{O2} = 1.239$  (5) Å] agree with values reported in the literature (Ocak Iskeleli *et al.*, 2005).

The crystal structure is stabilized by an intramolecular  $\text{N1}-\text{H1}\cdots\text{O2}$  hydrogen bond and two  $\pi-\pi$  stacking interactions. These interactions are between  $\text{Cg1}(\text{O1}-\text{C8})$  at  $(x, y, z)$  and  $\text{Cg1}$  at  $(1-x, -y, -z)$  [ $\text{Cg1}\cdots\text{Cg1}^i = 3.500$  (3) Å],  $\text{Cg3}(\text{C12}-\text{C17})$  at  $(x, y, z)$  and  $\text{Cg2}(\text{C1}-\text{C6})$  at  $(1-x, 1-y, -z)$  [ $\text{Cg3}\cdots\text{Cg2}^{ii} = 3.691$  (3) Å].

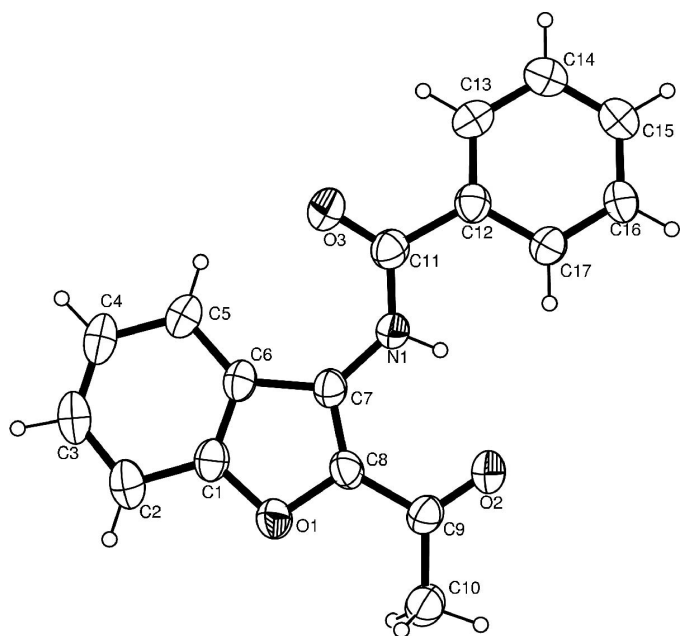
## Experimental

A mixture of 2-acetyl-3-aminobenzofuran (1.75 g, 10 mmol) and benzoyl chloride (1.54 g, 11 mmol) in absolute acetone (200 ml) was stirred at room temperature for 5 h. The reaction mixture was poured

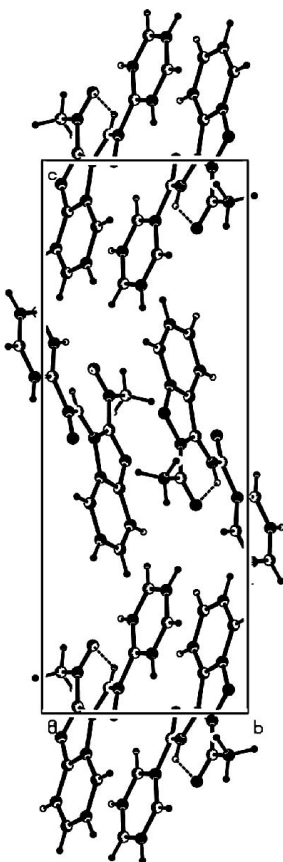
Received 12 April 2005

Accepted 25 April 2005

Online 7 May 2005



**Figure 1**  
The structure of (I), showing the atom-labelling scheme. Displacement ellipsoids are drawn at the 50% probability level and H atoms are shown as small spheres of fixed radius.



**Figure 2**  
A packing diagram of compound (I) viewed along the *a* axis.

into water (400 ml) and neutralized with ammonia (5%). The title compound was filtered off, washed with water, dried and recrystallized from acetone (86%) to yield colourless crystals.

**Crystal data**

$C_{17}H_{13}NO_3$   
 $M_r = 279.28$   
Monoclinic,  $P2_1/c$   
 $a = 8.9283$  (12) Å  
 $b = 7.4852$  (12) Å  
 $c = 20.273$  (3) Å  
 $\beta = 97.032$  (11)°  
 $V = 1344.7$  (3) Å<sup>3</sup>  
 $Z = 4$

$D_x = 1.380$  Mg m<sup>-3</sup>  
Mo  $K\alpha$  radiation  
Cell parameters from 7057 reflections  
 $\theta = 2.0$ – $26.9^\circ$   
 $\mu = 0.10$  mm<sup>-1</sup>  
 $T = 293$  (2) K  
Plate, colourless  
 $0.80 \times 0.34 \times 0.08$  mm

**Data collection**

Stoe IPDS-2 diffractometer  
 $\omega$  scans  
Absorption correction: by integration (*X-RED32*; Stoe & Cie, 2002)  
 $T_{min} = 0.953$ ,  $T_{max} = 0.994$   
7795 measured reflections

2615 independent reflections  
964 reflections with  $I > 2\sigma(I)$   
 $R_{int} = 0.141$   
 $\theta_{max} = 26.0^\circ$   
 $h = -11 \rightarrow 10$   
 $k = -9 \rightarrow 9$   
 $l = -24 \rightarrow 24$

**Refinement**

Refinement on  $F^2$   
 $R[F^2 > 2\sigma(F^2)] = 0.073$   
 $wR(F^2) = 0.195$   
 $S = 0.84$   
2615 reflections  
191 parameters  
H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0791P)^2]$   
where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{max} < 0.001$   
 $\Delta\rho_{max} = 0.21$  e Å<sup>-3</sup>  
 $\Delta\rho_{min} = -0.29$  e Å<sup>-3</sup>  
Extinction correction: *SHELXL97*  
Extinction coefficient: 0.015 (3)

**Table 1**

Selected bond lengths (Å).

C1—C6	1.361 (6)	C6—C7	1.464 (5)
C1—O1	1.382 (5)	C7—C8	1.347 (6)
C1—C2	1.415 (6)	C7—N1	1.376 (5)
C2—C3	1.383 (7)	C8—O1	1.412 (5)
C3—C4	1.346 (7)	C9—O2	1.239 (5)
C4—C5	1.405 (6)	C11—O3	1.232 (5)
C5—C6	1.399 (6)		

**Table 2**

Hydrogen-bonding geometry (Å, °).

<i>D</i> —H... <i>A</i>	<i>D</i> —H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> —H... <i>A</i>
N1—H1...O2	0.86	2.13	2.768 (5)	131

H atoms were positioned geometrically and treated using a riding model, with aromatic C—H distances of 0.93 Å, methyl C—H distances of 0.96 Å and an N—H distance of 0.86 Å.  $U_{iso}(H)$  values were set at  $xU_{eq}(\text{carrier atom})$ , where  $x = 1.5$  for methyl H and 1.2 for other H atoms. Owing to the poor quality of the crystal, the  $R_{int}$  and  $R$  values are somewhat high.

Data collection: *X-AREA* (Stoe & Cie, 2002); cell refinement: *X-AREA*; data reduction: *X-RED32*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEPIII* (Burnett & Johnson, 1996); software used to prepare material for publication: *WinGX* (Farrugia, 1999).

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